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Fuel chemistry and thermodynamic aspects under irradiation

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Thermodynamics and nuclear fuel



$$\Delta T(r) = T(R) - T(r) = \frac{\chi}{4\pi\lambda R^2} (R^2 - r^2)$$

- = linear heating rate (W cm⁻¹)
- = pellet radius (cm)

χ

R

λ

= Thermal conductivity (W cm⁻¹ K⁻¹)



Thermodynamic equilibrium?



- Restructuring takes place during the fist hours
- Massive matter transport
- Fission changes the composition
- Radiation



Thermodynamics and nuclear fuel





Cross section of a fast reactor mixed oxide fuel from the DS1 experiment





(a) $(Np_{0.016}Am_{0.016}Pu_{0.3}U_{0.668})O_{1.98}$ irradiated for 10 min at 427 W/cm



(b) $(Np_{0.016}Am_{0.016}Pu_{0.3}U_{0.668})O_{1.98}$ irradiated for 24 h at 432 W/cm



6

(c) $(Np_{0.016}Am_{0.016}Pu_{0.3}U_{0.668})O_{1.96}$ irradiated for 24 h at 429 W/cm

Short irradiation tests of minor actinide fuels performed in the JOYO reactor in Japan

Two different O/M ratios







Several transport processes take place

- Vaporisation-condensation
- Matrix diffusion of oxygen and metal
- Grain boundary diffusion















Maeda et al., J. Nucl. Mater. 2009, 389:78



11







O/M = 1.98

Am-MOX: 0.5 % Am

Np-Am-MOX: 2% Am+2% Np

Fig. 3a. Ceramographs of specimens sectioned from the fuel pin of O/M ratio 1.98 irradiated in the Am1-2 experiment.



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Fig. 3b. Ceramographs of specimens sectioned from the fuel pin of O/M ratio 1.95 irradiated in the Am1-2 experiment.

Alpha radiography



14 O'Boyle et al, 1969



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Alpha radiography



O'Boyle et al, 1969

15





• Results from the AIM1 irradiation in JOYO







Sari anmd Schumacher, JNM, 41:192



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SUPERFACT (U_{0.6}Np_{0.2}Am_{0.2})O₂ fuel

- Actinide burnout 31% (actinide fission not known)
- Swelling (axial expansion 2.3%, radial expansion 3.3%)
- High helium production & release (high porosity)





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- 1) Dissolved in the matrix: Rb, Sr, Y, Zr, Nb, Te, Cs, Ba, La, Ce, Pr, Nd, Pm, Sm, Eu
- 2) Oxide precipitates at grain boundaries: Rb, Sr, Zr, Nb, Mo, Se, Te, Cs, Ba
- 3) Metallic precipitates: Mo, Tc, Ru, Rh, Pd, Ag, Cd, In, Sn, Sb, Se, Te
- 4) Volatiles: Br, Rb, I, Cs, Te
- 5) Gases: Kr, Xe













- "Grey phase" adjacent to columnar grains
 - Perovskite-type phase
 (Ba,Sr,Cs)(Zr,U,Pu,Mo)O₃
- "White inclusions"
 - (Pd,Tc,Rh,Tc,Mo)

















X-ray mapping of the JOG





- Atomic diffusion in the lattice (thermal and radiation induced)
- 2. Capture in intergranular bubbles
- 3. Migration of bubbles to grain boundaries
- 4. Resolution of gas
- 5. Aggregation to closed porosity
- 6. Venting via open porosity channels





SEM of before KEMS SEM after heating at T = 1800 K in KEMS measurement Exp 7854 07070529 1800K innen n 7119 07030508, Mox , Ausg Mat

SE 7119 07030508, Mox Ausg Mat.

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Konings et al., In: The Chemistry of Actinides and Transactinide Elements, 4th Edition., Volume 6, Chapter 34, p. 3665-3812, Springer Netherlands, 2010



Cesium iodide

- Fragments: Cs⁺, I⁺, CsI⁺, Cs₂I⁺, Cs₂⁺, I₂⁺
- I⁺/CsI⁺ ≈ 3:1
- Parallel release profiles

Simulated Fuel (UO₂+CsI)

- Fragments: Cs⁺, I⁺, CsI⁺
- I⁺/CsI⁺ ≈ 3:2
- Parallel release profiles





32



Irradiated fuel (BWR, 55 MWd/kg)

- Absolute intensities similar to Simfuel
- No Csl⁺ ions (but higher background)
- Some temperature regions with parallell release

Conclusion: No clear evidence for CsI formation in irradiated fuel

- Insufficient reaction sites?
- Gamma radiation?



Summary:

- Nuclear fuel is not in chemical equilibrium
- But at microscopic scale processes can be described by equilibrium thermodynamics (local equilibrium)
- Chemical thermodynamics give the boundary conditions; kinetics need to be considered also



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